

CLINICAL SCIENCE

1988

Guidance for Authors

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1. POLICY OF THE JOURNAL

1.1. Scope

Clinical Science publishes papers in the field of clinical investigation, provided they are of a suitable standard and contribute to the advancement of knowledge in this field. The term 'clinical investigation' is used in its broadest sense to include studies in animals and the whole range of biochemical, physiological, immunological and other approaches that may have relevance to disease in man. Studies which are confined to normal subjects, or animals, or are purely methodological in nature may be acceptable. The material presented should permit conclusions to be drawn and should not be only of a preliminary nature. The journal publishes four types of manuscript, namely invited Editorial Reviews, Full Papers, Rapid Communications and Correspondence. In addition, *Clinical Science* publishes abstracts of the proceedings of the Medical Research Society (as Supplements) and also the Society's Annual Guest Lecture.

1.2. The Editorial Board

The Board comprises Editors for the Medical Research Society and the Biochemical Society and a Chairman and two Deputy Chairmen who are drawn alternately from the two Societies. Members of the Board retire after a maximum of 5 years; the Chairman serves in his capacity for 2 years. The membership of the Board is designed to cover as wide a range of interests as possible.

The main function of the Board is to decide on the acceptability of submitted papers, but it also deals with general matters of editorial policy. Financial policy is dealt with separately by the Committee of Management.

1.3. The editorial process

A submitted paper is first read by the Chairman or one of the Deputy Chairmen of the Editorial Board who then sends it to an Editor. The latter considers the paper in detail and sends it to one or more referees (who remain anonymous) from outside the membership of the Board. The Editor returns it with his recommendation to the Chairman who then writes formally to the authors. The ultimate responsibility of acceptance for publication lies with the Chairman. If the Chairman is for any reason unavailable, a Deputy Chairman assumes this function.

1.4. Ethics of investigations

(a) Human subjects. Authors must state in the text of their paper the manner in which they have complied,

where necessary, with the recommendations on human investigations published in the Medical Research Council report of 1962/63 [*British Medical Journal* (1964) ii, 178–180]. Consent **must** be obtained from each patient or subject after full explanation of the purpose, nature and risks of all procedures used, and the fact that such consent has been given should be recorded in the paper. Papers should also state that the Ethical Committee of the Institution in which the work was performed has given approval to the protocol.

(b) **Animals.** Care must always be taken to ensure that experimental animals do not suffer unnecessarily. Authors must state in the text the anaesthetic procedures used in full, and all precautions they took to ensure that the animals did not suffer unduly during and after the experimental procedure.

The Editorial Board will not accept papers where the ethical aspects are, in the Board's opinion, open to doubt.

1.5. Originality of papers

Submission of a paper to the Editorial Board is taken to imply that it reports work that has not been published in either the same or a substantially similar form, that it is not under consideration for publication elsewhere and that, if accepted for publication by *Clinical Science*, it will not be published elsewhere in the same form, either in English or in any other language, without the consent of the Editorial Board. This does not usually apply to previous publication of oral communications in brief abstract form. In such cases authors should enclose copies of the abstracts or previous publications. The author, or in the case of multiple authorship the authors, will be asked to sign a statement vesting the copyright in the publishers. Requests for consent for reproduction of material published in *Clinical Science* should be addressed to the Editorial Manager.

2. SUBMISSION OF MANUSCRIPTS: GENERAL INFORMATION AND FORMAT

2.1. General

Papers submitted for publication should be sent to the Editorial Manager, *Clinical Science*, 7 Warwick Court, London WC1R 5DP, U.K.

The submission should contain four copies (of which three may be photocopies) of the typescript, Tables, Figures etc. The authors should retain one copy of the paper. Typescripts produced on dot-matrix printers that are not of 'near letter quality' may be unacceptable. The Editorial Board does not accept responsibility for damage or loss of papers submitted, although great care is taken to ensure safety and confidentiality of the typescript during the editorial process. In the case of multiple authorship, the covering letter should indicate that the approval of all co-authors has been obtained and should be signed by all authors.

Papers should be presented so that they are intelligible to the non-specialist reader of the journal. This is particularly important in highly specialized fields and a very brief résumé of the current state of knowledge is usually help-

ful. Certain types of material, e.g. mathematical formulations requiring more than trivial derivations, should be given in a separate Appendix.

Where the reader is referred to previous works by the same author(s) for important details relevant to the present work, copies or reprints of the publication should be sent with the typescript. This is of particular importance in relation to methodology.

The dates of receipt and acceptance of the paper will be published. If the paper has to be returned to the authors for revision and is not resubmitted within 1 month, the date of receipt will be revised accordingly. Papers returned by authors later than 3 months after notification that revision is required will normally be treated as new submissions. It is emphasized that badly presented or unduly long papers will be returned for revision and delays in publication will be inevitable. Similar delays will be incurred if the typescript is not prepared strictly in accordance with the instructions detailed below.

Typescripts of rejected work will not be returned to authors unless a specific request for the return has been made at the time of submission.

2.2. Full papers

The authors should refer to a current issue of *Clinical Science* to make themselves familiar with the general layout. Concise presentation is very important for rising costs are a severe constraint on space. **The length of manuscript and the number of Figures and Tables must be kept to a minimum.** Extensive Tables of data can be deposited with the Royal Society of Medicine (see 2.5). *Guidance for Authors* is usually published in the January issue of the journal, and revised periodically.

Typescripts should be, in general, arranged as follows:

(a) *Title page.* Title: this should be as informative as possible, since titles of papers are being increasingly used in indexing and coding for information storage and retrieval. The title should indicate the species in which the observations reported have been made. The numbering of parts in a series of papers is not permitted.

List of authors' names (degrees and appointments are not required).

Laboratory or Institute of origin.

Key words: for indexing the subject of the paper; they should, if possible, be selected from the current issues of 'Medical Subject Headings' (MeSH), produced by the *Index Medicus*.

Short title: for use as a running heading in the printed text; it should not exceed forty-five characters and spaces and should not contain any abbreviations.

Author for correspondence: the name and address of the author to whom queries and requests for reprints should be sent.

(b) *Summary.* This should be a brief statement arranged in **numbered paragraphs** of what was done, what was found and what was concluded, and should rarely exceed 250 words. Abbreviations should be avoided as far as possible and must be defined. Statistical and methodological details including exact doses should also

be avoided unless they are essential to the understanding of the Summary.

(c) *Introduction*. This should be comprehensible to the general reader and should contain a clear statement of the reason for doing the work, but should not include either the findings or the conclusions.

(d) *Methods*. The aim should be to give sufficient information in the text or by reference to permit the work to be repeated without the need to communicate with the author.

(e) *Results*. This section should not include material appropriate to the Discussion section.

(f) *Discussion*. This should not contain results and should be pertinent to the data presented.

(g) *Acknowledgments*. These should be as brief as possible.

(h) *References*. See p. v for the correct format.

(i) *Figures and Tables*. See p. iv.

2.3. Rapid Communications

The passage of these papers through the editorial process will be expedited and contributors are encouraged to take advantage of this facility when rapid publication is of importance and when material can be presented concisely. Rapid Communications should describe completed work and should not be merely a preliminary communication.

To achieve rapid publication only a decision will be given, not an editorial report. In addition, authors of accepted Rapid Communications will not be sent proofs. Rejection of a paper submitted as a Rapid Communication does not preclude its re-submission as a full paper for publication in *Clinical Science*, in which event the paper would be reviewed and reports provided with the editorial decision in the normal way.

Papers intended for rapid publication should conform to the following criteria. The format should be similar to that for a full paper. The text should not exceed 1800 words. One Figure or Table is allowed; if neither is included then the text may be expanded to 2400 words.

2.4. Correspondence

Letters containing original observations or critical assessments of material published in *Clinical Science*, including Editorial Reviews, will be considered for the Correspondence section of the journal. Letters should be no longer than 750 words, with one Figure or Table and up to six references, or 1000 words maximum without a Figure or Table. Letters relating to material previously published in *Clinical Science* should be submitted within 6 months of the appearance of the article concerned. They will be sent to the authors for comment and both the letter and any reply by the author will be published together. Further correspondence arising therefrom will also be considered for publication. Consideration will also be given to publication of letters on ethical matters.

2.5. Arrangements for large amounts of information

It is impracticable to publish very large sets of individual values or very large numbers of diagrams, and

under these circumstances a summary of the information only should be included in the paper. The information from which the summary was derived should be submitted with the typescript and, if the latter is accepted, the Editors may ask for a copy of the full information and diagrams to be deposited with the Librarian, the Royal Society of Medicine, 1 Wimpole Street, London W1M 8AE, who will issue copies on request. Experience has shown that such requests are frequently received.

2.6. Proof corrections

These are expensive and corrections of other than printers' errors may have to be charged to the author.

2.7. Offprints

Fifty offprints are supplied free and additional copies may be obtained at terms, based upon the cost of production, that will be given with the proofs. All offprints should be ordered when the proofs are returned.

2.8. Availability on MEDLINE

Summaries of papers in *Clinical Science* are available on-line on teleprinters participating in the MEDLINE system run by the National Library of Medicine, National Institutes of Health, Bethesda, Maryland, U.S.A.

3. MISCELLANEOUS NOTES

3.1. Abbreviations

Abbreviations should be avoided; if used they must be defined at the first mention; new abbreviations should be coined only for unwieldy names which occur frequently. Abbreviations should not appear in the title nor, if possible, in the Summary. A list of accepted abbreviations is on p. vi. Numbers, not initials, should be used for patients and subjects.

3.2. Anatomical nomenclature

This should follow the recommendations of the International Anatomical Nomenclature Committee (1966) *Nomina Anatomica*, 3rd edn, Excerpta Medica Foundation, Amsterdam.

3.3. Animals, plants and micro-organisms

The full binomial specific names should be given at first mention for all experimental animals other than common laboratory animals. The strain and, if possible, the source of laboratory animals should be stated. Thereafter in the text, single letter abbreviations may be given for the genus; if two genera with the same initial letter are studied, abbreviations such as *Staph.* and *Strep.* should be used.

3.4. Buffers and salts

The acidic and basic components should be given, together with the pH. Alternatively, a reference to the composition of the buffer should be given. Further details are provided in the *Biochemical Journal* (1988) **249**, 9.

When describing solutions containing organic anions and their parent acids, the salt designator (e.g. lactate,

urate, oxalate) should be used in preference to the name of the acid (lactic, uric, oxalic) unless it is certain that virtually all of the acid is in the undissociated form.

The composition of incubation media should be described, or a reference to the composition should be given.

3.5. Computer modelling

Papers concerned primarily with computer modelling techniques are acceptable provided that use of such techniques leads to a clear choice between two or more alternative hypotheses, or to the formulation of a new hypothesis amenable to experimental challenge or verification, or provides some new insight into the behaviour of a particular physiological system. Extensive technical details of hardware and software should not be given.

3.6. Doses

Doses of drugs should be expressed in mass terms, e.g. milligrams (mg) or grams (g), and also (in parentheses) in molar terms, e.g. mmol, mol, where this appears to be relevant. Molecular weights of many drugs may be found in *The Merck Index*, 10th edn, Merck and Co. Inc., N.J., U.S.A.

3.7. Enzymes

Nomenclature should follow that given in *Enzyme Nomenclature* (1984), Academic Press, London and New York, and the Enzyme Commission (EC) number should be quoted at the first mention. Where an enzyme has a commonly used informal name, this may be employed after the first formal identification. A unit of enzyme activity should preferably be expressed as that amount of material which will catalyse transformation of 1 μmol of the substrate/s under defined conditions, including temperature and pH. This gives the unit of the amount of enzyme named the katal (symbol kat). Alternatively, or when the natural substrate has not been fully defined, activity should be expressed in terms of units of activity relative to that of a recognized reference preparation, assayed under identical conditions. Activities of enzymes should normally be expressed as units/ml or units/mg of protein.

3.8. Evaluation of measurement procedures

When a new measuring procedure has been used, or when an established procedure has been applied in a novel fashion, an estimate of the precision of the procedure should be given. This should, as far as possible, indicate what sources of variation have been included in this estimate, e.g. variation of immediate replication, variation within different times of day, or from day to day etc.

If the precision of measurement varies in proportion to the magnitude of the values obtained, it can best be expressed as the coefficient of variation; otherwise it should be expressed by an estimate of the (constant) standard error of a single observation, or by estimates of several points within the range of observed values.

When recovery experiments are described the approximate ratio of the amount added to the amount already

present and the stage of the procedure at which the addition was made should be stated.

3.9. Figures and Tables

These are expensive to print and their number should be kept to a minimum. Their appropriate position in the paper should be indicated in the margin of the text. References to Figures and Tables should be in Arabic numerals, e.g. Fig. 3, and they should be numbered in order of appearance. In general, the same data should not be presented in both a Figure and a Table.

Figures, with captions attached, should be supplied as original drawings or photographs together with photocopies. All Figures should have their number and the authors' names written in pencil on the back; the top of the Figure should be indicated with a pencilled arrow. Acceptable symbols for experimental points are \bullet , \blacktriangle , \blacksquare , \circ , \triangle , \square . The symbols \times or $+$ must be avoided. The same symbols must not be used for two curves where the points might be confused. For scatter diagrams, solid symbols are preferred. When a particular variable appears in more than one Figure, the same symbol should be used for it throughout, if possible.

Curves should not be drawn beyond the experimental points, nor should axes extend appreciably beyond the data. Only essential information that cannot readily be included in the legend should be written within the Figure.

Figures for half-tone reproduction should be submitted as glossy prints. They are particularly expensive to print and their use should be avoided as far as possible.

Tables should be typed separately from the text. They should have an underlined title followed by any legend.

Captions for the Figures, and titles and legends for the Tables, should make then **readily understandable** without reference to the text. Adequate statistical information, including that on regression lines, should be included in Figure captions where appropriate.

3.10. Footnotes

These should be avoided as far as possible but where they are used in Tables they should be identified by the symbols * † ‡ § || ¶, in that order.

3.11. Isotope measurements

The information given should include (a) conditions of radioactivity counting, e.g. infinitely thick, infinitely thin; (b) the nature of the phosphor used in liquid-scintillation counting; (c) details of corrections made to the observed count rate, e.g. for 'quenching' or 'cross-over'; (d) standard deviation of the results or a statement of the *minimum total counts above background collected and the background value.*

In general the specific radioactivity of the starting materials should be given, preferably in terms of radioactivity per unit weight or, for stable isotopes, as atoms % excess.

Pending the general introduction of SI units radioactivity should continue to be expressed in terms of the curie (Ci) followed by the corresponding figure in

terms of the becquerel (Bq: disintegrations/s), in parentheses, and suitably rounded.

3.12. Radionuclide applications in man

If new or modified radionuclide applications in man are described, an estimate of the maximal possible radiation dose to the body and critical organs should be given.

For the time being this can continue to be expressed in rems, but with the corresponding figure in sieverts (Sv) given in parentheses after it.

3.13. Methods

In describing certain techniques, namely centrifugation (when the conditions are critical), chromatography and electrophoresis, authors should follow the recommendations published by the Biochemical Society [currently, *Biochemical Journal* (1988) **249**, 1–20].

3.14. Nomenclature of disease

This should follow the *International Classification of Disease* (8th revision, World Health Organization, Geneva, 1969) as far as possible.

3.15. Powers in Tables and Figures

Care is needed where powers are used in Table headings and in Figures to avoid numbers with an inconvenient number of digits. For example: (i) an entry '2' under the heading $10^3 k$ means that the value of k is 0.002; an entry '2' under the heading $10^{-3} k$ means that the value of k is 2000. (ii) A concentration 0.00015 mol/l may be expressed as 0.15 under the heading 'concn. (mmol/l)' or as 150 under the heading 'concn. ($\mu\text{mol/l}$)' or as 15 under the heading ' $10^5 \times \text{concn. (mol/l)}$ ', but not as 15 under the heading 'concn. (mol/l $\times 10^{-5}$)'.

3.16. References

The numerical citation system is used: references in the text are numbered consecutively in the order in which they are first mentioned, the numerals being given in brackets, e.g. [22]. References cited in Figure legends or Tables only should be numbered in a sequence determined by the position of the first mention in the text of the Figure or Table. References should be listed in numerical order and the names of all authors of a paper should be given, with the full title of the paper and the source details in full including the first and last page numbers, e.g.

2. Clark, T.J.H., Freedman, S., Campbell, E.J.M. & Winn, B.R. (1969) The ventilatory capacity of patients with chronic airways obstruction. *Clinical Science*, **36**, 307–316.

When the quotation is from a book, the following format should be used, giving the relevant page or chapter number:

20. Mollison, P.L. (1967) *Blood Transfusion in Clinical Medicine*, 4th edn, p. 50. Blackwell Scientific Publications, Oxford.
22. Reid, L. (1968) In: *The Lung*, p. 87. Ed. Liebow, A.A. & Smith, D.E. Williams and Wilkins, Baltimore.

References to 'personal communications' and 'unpublished work' should appear in the text only and not in

the list of references. The name and initials of the source of information should be given. When the reference is to material that has been accepted for publication but has not yet been published, this should be indicated in the list of references by 'In press' together with the name of the relevant journal and, if possible, the expected date of publication. If such a citation is of major relevance to the manuscript submitted for publication authors are advised that the editorial process might be expedited by the inclusion of a copy of such work. In the case of quotations from personal communications the authors should state in the covering letter that permission for quotation has been obtained.

3.17. Solutions

Concentration of solutions should be described where possible in molar terms (mol/l and subunits thereof), stating the molecular particle weight if necessary. Values should not be expressed in terms of normality or equivalents. Mass concentration should be expressed as g/l or subunits thereof, for example mg/l or $\mu\text{g/l}$. For solutions of salts, molar concentration is always preferred to avoid ambiguity as to whether anhydrous or hydrated compounds are used. Concentrations of aqueous solutions should be given as mol/l or mol/kg (g/l or g/kg if not expressed in molar terms) rather than % (w/v) or % (w/w). It should always be made clear whether concentrations of compounds in a reaction mixture are final concentrations or the concentrations in solutions added.

3.18. Spectrophotometric data

The term 'absorbance' [$\log(I_0/I)$] should be used rather than 'optical density' or 'extinction'. The solvent, if other than water, should be specified. Symbols used are: A , absorbance; a , specific absorption coefficient (litre $\text{g}^{-1} \text{cm}^{-1}$) (alternatively use $A_{1\%}^{1\text{cm}}$); ϵ , molar absorption coefficient (the absorbance of a molar solution in a 1 cm light-path) (litre $\text{mol}^{-1} \text{cm}^{-1}$, not $\text{cm}^2 \text{mol}^{-1}$).

3.19. Spelling

Clinical Science uses as standards for spelling the *Concise* or *Shorter Oxford Dictionary of Current English* (Clarendon Press, Oxford) and *Butterworth's Medical Dictionary* (Butterworths, London).

3.20. Statistics

Papers are frequently returned for revision (and their publication consequently delayed) because the authors use inappropriate statistical methods. Two common errors are the use of means, standard deviations and standard errors in the description and interpretation of grossly non-normally distributed data and the application of t -tests for the significance of difference between means in similar circumstances, or when the variances of the two groups are non-homogeneous. In some circumstances it may be more appropriate to provide a 'scattergram' than a statistical summary. Authors are recommended to consult the statistical guidelines presented by Altman *et al.* in 'Statistical guidelines for contributors to medical journals' *British Medical Journal* (1983) **286**, 1489–1493.

The type of statistical test used should be stated in the Methods section. A reference should be given for the less commonly encountered statistical tests. The format for expressing mean values and standard deviations or standard errors of the mean is, for example: mean cardiac output 10.4 l/min (SD 1.2; $n=11$). Degrees of freedom should be indicated where appropriate. Levels of significance are expressed in the form $P < 0.01$.

3.21. Trade names

The name and address of the supplier of special apparatus and of biochemicals should be given. In the case of drugs, approved names should always be given with trade names and manufacturers in parentheses.

4. UNITS: THE SI SYSTEM

The recommended *Système International* (SI) units [see *Quantities, Units and Symbols*, 2nd edn (1975) The Royal Society, London] are used by *Clinical Science*. **All papers submitted should use these units** except for blood pressure values, which should be expressed in mmHg, or gas tensions, where values at the author's discretion may be given in mmHg (with kPa in parentheses) or as kPa (with mmHg in parentheses) in the text and either as mmHg or as kPa in Figures, which (if practicable) should have scales in both units. Airways pressure should be expressed in kPa. Where molecular weight is known, the amount of a chemical or drug should be expressed in mol or in an appropriate subunit, e.g. mmol. Energy should be expressed in joules (J).

The basic SI units and their symbols are as follows:

Physical quantity	Name	Symbol
length	metre	m
mass	kilogram	kg
time	second	s
electric current	ampere	A
thermodynamic temperature	kelvin	K
luminous intensity	candela	cd
amounts of substance	mole	mol

The following are examples of derived SI units:

Physical quantity	Name	Symbol	Definition
energy	joule	J	$\text{kg m}^2 \text{s}^{-2}$
force	newton	N	$\text{kg m s}^{-2} = \text{J m}^{-1}$
power	watt	W	$\text{kg m}^2 \text{s}^{-3} = \text{J s}^{-1}$
pressure	pascal	Pa	$\text{kg m}^{-1} \text{s}^{-2} = \text{N m}^{-2}$
electric charge	coulomb	C	A s
electric potential difference	volt	V	$\text{kg m}^2 \text{s}^{-2} \text{A}^{-1} = \text{J A}^{-1} \text{s}^{-1}$
electric resistance	ohm	Ω	$\text{kg m}^2 \text{s}^{-3} \text{A}^{-2} = \text{V A}^{-1}$
electric conductance	siemens	S	$\text{kg}^{-1} \text{m}^{-2} \text{s}^3 \text{A}^2 = \Omega^{-1}$
electric capacitance	farad	F	$\text{A}^2 \text{s}^3 \text{kg}^{-1} \text{m}^{-2} = \text{A s V}^{-1}$
frequency	hertz	Hz	s^{-1}
volume	litre	l	10^{-3}m^3

The word 'litre' has been accepted as a special name for cubic decimetre (1 litre = 1 dm^3).

Both the basic and derived SI units, including the symbols of derived units that have special names, may be preceded by prefixes to indicate multiples and sub-multiples. The prefixes should be as follows:

Multiple	Prefix	Symbol	Multiple	Prefix	Symbol
10^6	mega	M	10^{-3}	milli	m
10^3	kilo	k	10^{-6}	micro	μ
10^2	hecto	h*	10^{-9}	nano	n
10	deka	da	10^{-12}	pico	p
10^{-1}	deci	d*	10^{-15}	femto	f
10^{-2}	centi	c*			

*To be avoided where possible (except for cm).

Compound prefixes should not be used, e.g. 10^{-9} m should be represented by 1 nm, not 1 m μm .

Notes:

- (i) Full stops are not used after symbols.
- (ii) Minutes (min), hours (h), days and years will continue to be used in addition to the SI unit of time [the second (s)].
- (iii) The solidus may be used in a unit as long as it does not have to be employed more than once, e.g. mmol/l is acceptable, but ml/min/kg is not, and should be replaced by $\text{ml min}^{-1} \text{kg}^{-1}$.

5. ABBREVIATIONS, CONVENTIONS, DEFINITIONS, SYMBOLS AND SPECIAL COMMENTS

As well as standard symbols and abbreviations that have been accepted by international bodies, and which can be used without definition, this list shows selected abbreviations in the form of groups of capital letters (e.g. ALA, ECF, MCHC) which when used must be defined in the text as indicated on p. iii. The standard abbreviations for amino acids are only for use in Figures and Tables or for peptide sequences.

absorbance	A
acceleration due to gravity	g
adenosine 3':5'-cyclic monophosphate	cyclic AMP
adenosine 5'-phosphate	AMP
adenosine 5'-pyrophosphate	ADP
adenosine triphosphatase	ATPase
adenosine 5'-triphosphate	ATP
adrenoceptor (<i>see also</i> blocking agents)	
adrenocorticotrophic hormone	ACTH
alanine	Ala
alternating current	a.c.
alveolar minute ventilation	\dot{V}_A
alveolar to arterial oxygen tension difference	$(P_{A\text{O}_2} - P_{a\text{O}_2})$
aminolaevulinic acid	ALA
ampere	A

angiotensin	ANG; reference amino acid abbreviations are used as prefix within brackets: e.g. [Sar ¹ , Val ⁵ , Ala ⁸]ANG	disintegrations/min disintegrations/s dissociation constant	d.p.m. d.p.s.
Ångstrom (Å)	<i>not used</i> ; express in nm (1 Ångstrom = 10 ⁻¹ nm)	acidic apparent basic minus log of	K_a e.g. K'_a K_b pK
antidiuretic hormone	ADH (when referring to the physiological secretion)	doses	avoid Latin designations such as b.d. and t.i.d.
arginine	Arg	dyne	dyn; used for vascular resistance
arteriovenous	a-v: <i>permitted</i> in Figures and Tables	elastance	E ; express in Pa m ⁻³
asparagine	Asn	electrocardiogram	ECG
aspartic acid	Asp	electroencephalogram	EEG
atmosphere (unit of pressure)	<i>not used</i> ; express in kPa (1 atmosphere = 101.325 kPa)	electromotive force	e.m.f.
atomic weight	at. wt.	electron spin resonance	e.s.r.
becquerel	Bq (1 d.p.s.)	electronvolt	eV (for radiation energies)
blocking agents	e.g. β -adrenoceptor antagonists preferred	equation	eqn.
blood pressure	express in mmHg	equivalents (amount of a chemical)	<i>not used</i> ; recalculate in molar terms
blood urea nitrogen	<i>not used</i> ; recalculate as urea, express in mmol/l	erythrocyte count	express as 10 ¹² cells/l
blood volume	BV	erythrocyte sedimentation rate	ESR
body temperature and pressure, saturated	BTPS	ethanol, ethanolic	not ethyl alcohol or alcoholic
British Pharmacopoeia	write in full and give edition	ethylenediaminetetra-acetate	EDTA
calculated	calc. (in Tables only)	'ethyleneglycolbis(aminoethyl-ether)'tetra-acetate'	EGTA
'Calorie' (= 1000 cal)	<i>not used</i> ; recalculate as kilojoules (1 'Calorie' = 4.184 kJ)	exchangeable	Na _e , K _e etc. for total exchangeable sodium, potassium etc.
carbon dioxide output (in respiratory physiology)	\dot{V}_{CO_2} ; express in ml STP/min	Experiment (with reference numeral)	Expt.; plural, Expts.
cardiac frequency	f_c ; in beats/min	expired minute ventilation	\dot{V}_E
cardiac output	express in l/min	extinction	<i>use</i> absorbance
centimetre	cm	extracellular fluid	ECF
clearance of x	C_x	extracellular fluid volume	ECFV
coenzyme A and its acyl derivatives	CoA and acyl-CoA	extraction ratio of x (renal)	E_x
compare	cf.	Figure (with reference numeral)	Fig.; plural, Figs.
complement fractions	C1-C9	filtered load of x (renal)	F_x
compliance (respiratory physiology)	C; express in l kPa ⁻¹	follicle-stimulating hormone	FSH
concentrated	conc.	forced expiratory volume in 1.0 s	FEV _{1.0}
concentration	concn.; may be denoted []; e.g. plasma [HCO ₃]	fractional concentration in dry gas	F
conductance (respiratory physiology)	G; express in l s ⁻¹ kPa ⁻¹	fractional disappearance rate	k (as in $A = A_0 e^{-kt}$)
correlation coefficient	r : may be used without definition	frequency of respiration	f_R ; in breaths/min
counts/min, counts/s	c.p.m., c.p.s.	functional residual capacity	FRC
cubic centimetres	<i>use</i> ml	gas-liquid chromatography	g.l.c.
curie	Ci (1 Ci = 3.7 × 10 ¹⁰ d.p.s.)	gas transfer factor	T ; in mmol min ⁻¹ kPa ⁻¹
cycle/s	Hz	glomerular filtration rate	GFR
cysteine	Cys	glutamic acid	Glu
dates	e.g. 11 August 1970	glutamine	Gln
dead-space minute ventilation	\dot{V}_D	glutathione	GSH (reduced); GSSG (oxidized)
dead-space volume	V_D	glycine	Gly
degrees, Celsius or centigrade	°C	gram(me)	g
deoxy (prefix)	<i>not</i> desoxy	gravitational field, unit of (9.81 m s ⁻¹)	g
deoxycorticosterone	DOC	gray	Gy (100 rads)
deoxycorticosterone acetate	DOCA	growth hormone	GH; if human, hGH
deoxyribonucleic acid	DNA	haematocrit	not allowed; <i>use</i> packed cell volume (PCV)
dialysate	diffusate preferred; 'dialysate' should be clearly defined	haemoglobin	Hb; express in g/dl
diethylaminoethylcellulose	DEAE-cellulose	half-life	$t_{1/2}$
differential of x with respect to time	\dot{x} (= dx/dt)	hertz (s ⁻¹)	Hz
1,25-dihydroxycholecalciferol	1,25-(OH) ₂ D ₃	high-pressure (or high-performance) liquid chromatography	h.p.l.c.
dilute	dil.	histidine	His
2,3-diphosphoglycerate	2,3-DPG	hour	h
direct current	d.c.	human chorionic gonadotropin	hCG
		human placental lactogen	hPL
		hydrocortisone	<i>use</i> cortisol
		hydrogen ion activity	aH; express in nmol/l
		minus log of	pH

25-hydroxycholecalciferol	25-(OH) ₂ D ₃	nicotinamide-adenine	NAD if oxidation state not
hydroxyproline	Hyp	dinucleotide	indicated
immunoglobulins	IgA, IgD, IgE, IgG, IgM		NAD ⁺ if oxidized
injection routes:	use abbreviations only in		NADH if reduced
intra-arterial	Figures	nicotinamide-adenine	NADP if oxidation state not
intramuscular	i.a.	dinucleotide phosphate	indicated
intraperitoneal	i.m.		NADP ⁺ if oxidized
intravenous	i.p.	normal	NADPH if reduced
subcutaneous	i.v.		should not be used to denote
international unit	s.c.		the concentration or osmo-
	i.u. (definition and reference	normal temperature and	larity of a solution
	should be given for un-	pressure	use standard temperature and
	common or ambiguous	nuclear magnetic resonance	pressure (STP)
	applications, e.g. enzymes)	number (in enumerations)	n.m.r.
intracellular fluid	ICF	observed	no. (in Tables only)
intracellular fluid volume	ICFV	ohm	obs. (in Tables only)
ionic strength	<i>I</i>	ornithine	Ω
isoleucine	Ile	<i>ortho</i> -	Orn
isotonic	<i>not used</i> ; specify composition of	orthophosphate (inorganic)	<i>o</i> -
	fluid, e.g. NaCl, 150 mmol/l	osmolarity	P _i
isotopically labelled com-	e.g. [U- ¹⁴ C]glucose, [1- ¹⁴ C]glu-	oxygen uptake per minute	express in osmol (or mosmol)/l
pounds	cose, sodium [1- ¹⁴ C]acetate;	(in respiratory physiology)	V _{O₂} ; express in ml
	use ¹³¹ I-labelled albumin, <i>not</i>	packed cell volume	STP/min
	[¹³¹ I]albumin	page, pages	PCV
	for simple molecules: ¹⁴ CO ₂ ,	<i>para</i> -	p., pp.
	³ H ₂ O	<i>para</i> -aminohippurate	<i>p</i> -
joule	J	partial pressure	PAH
katal	kat		<i>P</i> ; express in either kPa or
kilogram(me)	kg	e.g. alveolar, of O ₂	mmHg (see p. 000)
kilopond	<i>not used</i> ; 1 kilopond =	arterial, of CO ₂	P _{AO₂}
	9.8067 N	capillary, of O ₂	P _{ACO₂}
lactate dehydrogenase	LDH	end-tidal, of CO ₂	P _{capO₂}
leucine	Leu	mixed venous, of CO ₂	P _{ETCO₂}
leucocyte count	express as 10 ⁹ cells/l		P _{VCO₂}
lipoproteins (serum)		pascal	Pa
high density	HDL	per	/
low density	LDL	per cent	%
very low density	VLDL	petroleum ether	<i>not used</i> ; use light petroleum
litre	l (write in full if confusion with		and give boiling range
	the numeral 1 is possible)	phenylalanine	Phe
logarithm (base 10)	log	plasma renin activity	express as pmol of angiotensin I
logarithm (base e)	ln		h ⁻¹ ml ⁻¹
luteinizing hormone	LH	plasma volume	PV
lysine	Lys	poise	1 poise = 10 ⁻¹ N s m ⁻²
maximum	max.	polyacrylamide-gel electro-	PAGE
mean corpuscular haemoglobin	MCH; express in pg	phoresis	
mean corpuscular haemoglobin	MCHC; express in g/dl	potential difference	p.d.
concentration		power output	W (1 kpm/min = 0.1635 W)
mean corpuscular volume	MCV; express in fl (1 μm ³ = 1 fl)	precipitate	ppt.
median lethal dose	LD ₅₀	pressure	<i>P</i> ; express in kPa (except for
melting point	m.p.		blood pressures and gas
<i>meta</i> -	<i>m</i> -	probability of an event being	tensions: see p. vi); 1 kPa =
methanol, methanolic	<i>not</i> methyl alcohol	due to chance alone	7.5 mmHg
methionine	Met	proline	<i>P</i>
metre	m	protein-bound iodine (plasma)	Pro
Michaelis constant	K _m	pulmonary capillary blood flow	PBI
micromole	μmol	pyrophosphate (inorganic)	<i>Q</i> _c
micron (10 ⁻⁶ m)	μm; <i>not</i> μ	rad (radiation dose; 10 ⁻⁵ J	PP _i
milliequivalent	<i>not used</i> ; give amount in mmol	absorbed/g of material)	not abbreviated
millilitre	ml	red blood cell	(100 rads = 1 Gy)
millimetre of mercury	mmHg; for blood pressure and,		use erythrocyte; express counts
	at authors' discretion, for gas	relative band speed (partition	as 10 ¹² cells/l
	tensions: see p. vi	chromatography)	R _f
	(1 mmHg = 0.133 kPa)	rem	100 ergs/s × quality factor
millimolar (concentration)	mmol/l; <i>not</i> mM	renin	see plasma renin activity
millimole	mmol	residual volume	RV
minimum	min.	resistance (rheological)	<i>R</i> ; express in kPa l ⁻¹ s
minute (60 s)	min	respiratory exchange	<i>R</i>
molal	mol/kg	ratio (pulmonary)	
molar (concentration)	mol/l; <i>not</i> M	respiratory quotient	RQ
molar absorption coefficient	ε (the absorbance of a molar	(metabolic)	
	solution in a 1 cm light-path)	revolutions	rev.
mole	mol		
molecular weight	mol. wt.		

rev./min	<i>not</i> r.p.m.; <i>use g</i> if possible (see p. vii)	thyrotrophin-releasing hormone	TRH
ribonucleic acid	RNA	tidal volume	V_T
röntgen	R	time (symbol)	t
saline	define at first mention [e.g. NaCl solution (154 mmol/l)]	time of day	e.g. 18.15 hours
saturation	S , e.g. S_{aO_2} for arterial oxygen saturation (see partial pressure for other analogous abbreviations)	total lung capacity	TLC
second (time)	s	tryptophan	Trp
serine	Ser	tubular maximal reabsorptive capacity for x	$T_{m,x}$
sievert	Sv (1 J/kg \times quality factor)	tyrosine	Tyr
solvent systems	e.g. butanol/acetic acid/water (4:1:1, by vol.), butanol/acetic acid (4:1, v/v)	ultraviolet	u.v.
► sodium dodecyl sulphate species	SDS	urinary concentration of x	U_x
specific activity	sp., plural spp. sp. act. Confusion must be avoided between e.g. specific radioactivity and the specific activity of an enzyme	valency	e.g. Fe^{2+} , <i>not</i> Fe^{++}
specific conductance of airways	sGaw; express in $s^{-1} kPa^{-1}$	valine	Val
standard deviation	SD } may be used	variance ratio	F
standard error of the mean	SEM } without definition	vascular resistance	express in $kPa l^{-1} s$ (with value in $dyn s cm^{-5}$ in parentheses); primary values of differential vascular pressure (mmHg) and flow (l/min) should always also be given in Tables or text as appropriate
standard temperature and pressure	STP	velocity	v , express as $m s^{-1}$
steroid nomenclature	see <i>Biochemical Journal</i> (1969) 113, 5–28; (1972) 127, 613–617	venous admixture	\dot{Q}_{va}
sulphydryl	<i>use</i> thiol or SH	veronal	used only for buffer mixtures; otherwise use 5,5'-diethylbarbituric acid
sum	Σ	viscosity, dynamic	η
Svedberg unit	S	viscosity, kinematic	ν
temperature (absolute)	T	vital capacity	VC
temperature (empirical)	t	volt	V
temperature, thermodynamic	K	volume of blood (in cardio-respiratory physiology)	Q ; <i>use</i> \dot{Q} for blood flow rate
thin-layer chromatography	t.l.c.	watt	W
threonine	Thr	wavelength	λ
thyrotrophic hormone	TSH	weight	wt.
		white blood cell	<i>use</i> leucocyte; express counts as 10^9 cells/l